

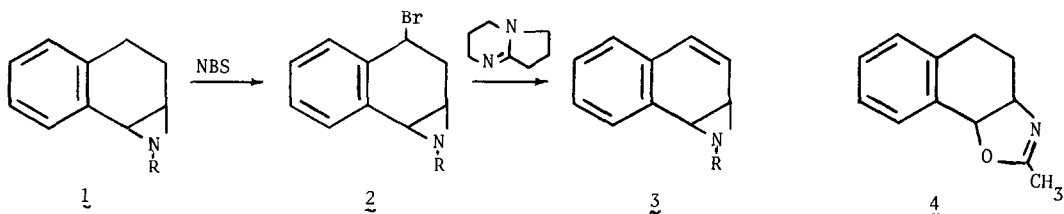
POLYCYCLIC ARENE IMINES

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Polycyclic arene oxides have been the focus of considerable research in recent years owing to their theoretical and biological relevance¹ Of similar interest,² but yet unreported, are polycyclic arene mono-imines Recent findings by Bicker and Fischer³ imply that some of the title compounds may play a similar part in chemical carcinogenesis to that of arene oxides

Previous attempts to prepare polycyclic mono-imines were unsuccessful⁴⁻⁶ Arene imines were proposed to be reaction intermediates in the syntheses of aryl carbamates from ethyl azidoformate and naphthalene,⁴ anthracene,^{5,6} phenanthrene⁵ and pyrene⁵, but proved impossible to isolate Under mild conditions naphthalene is reported to give some tetrahydronaphtho[1,2-b,3,4-b']bisazirine free of any monoaziridine derivative⁷

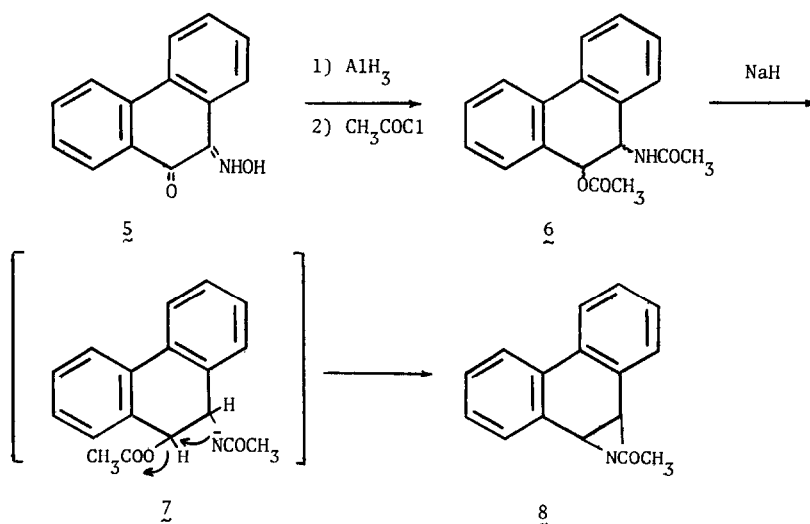
We now wish to report two general syntheses of polycyclic arene imines one which might be particularly useful for the introduction of an epimine function in a terminal ring, and one for preparation of so-called K region epimines



1a,2,3,7b-Tetrahydro-1H-naph[1,2-b]azirine (1, R=H) (prepared according to Heathcock and Hassner⁸) was acetylated at 0° with acetyl chloride in the presence of Et₃N to give 1, R=COCH₃ in quantitative yield⁹, bp 84° (0.02 mm)¹⁰, ν (C=O) 1675 cm⁻¹, ν (CH₂CN) 228 m μ (ϵ 7200) Subsequent bromination (NBS) afforded 2, R=COCH₃, δ (CCl₄) 5.46 (t, 1H, CHBr), mass spectrum (70 eV) 269, 267, 223, 225, 185, 183, 128 (base peak) The crude bromide (admixed with some oxazoline 4) was dehydrobrominated by stirring a THF solution (10 mmoles in 30 ml) with 1,5-diazabicyclo[4.3.0]nona-5-ene (2 ml) for 24 hr at -30 to -10° to yield 25% of

1-acetyl-1a,7b-dihydronaph[1,2-b]azirine (3, R=COCH₃) as a colorless oil. Mass spectrum (70 ev) 185, 183, 142, 128 (base peak), IR (C=O) 1665 cm⁻¹, no NH absorption appears in the IR spectrum, UV (CH₃CN) 242 mμ (ε 10800). The NMR spectrum resembles that of oxireno[a]naphthalene¹¹ δ (CCl₄) 1.46 (s, 3H), 4.83 (1H, J_{-1a,7b}=8 cps, J_{-1a,2}=4 cps, J_{-1a,3}=1.8 cps), 5.51 (d, 1H, J_{-1a,7a}=8 cps), the vinylic protons form an ABX pattern 5.82 (d, 1H, J_{-2,3}=10 cps, J_{-1a,2}=4 cps), 6.30 (d,d, 1H, J_{-2,3}=10 cps, J_{-1a,3}=1.8 cps), 7.03-7.34 (m, 4H). These data exclude the possibility of either ring cleavage of the aziridine structure to give 1- and 2-naphthylamine derivatives, or its expansion to the corresponding acetylbenzazepines. The imine 3, R=COCH₃ reacts readily with 1,3-diphenylisobenzofuran.

The method of preparation of K-substituted arene imines was demonstrated by the synthesis of 1-acetyl-1a,9c-dihydrophenanthr[9,10-b]azirine (8).



9,10-Phenanthrenequinone monoxime (5) was reduced with the required amount of a 2M solution of aluminium hydride in THF¹². Acetylation of the amino-alcohol, so formed, led to compound 6⁹, IR (Nujol) 1645, 1732 cm⁻¹ (amide and ester CO, respectively). Ring closure to 8 was afforded in 64% yield by addition of a suspension of 6 (10 mmoles in 150 ml benzene) to an equimolar amount of NaH in the same solvent (200 ml) under N₂ followed by reflux for 30 min. Alternatively 8 was obtained in < 10% yield when a THF solution of 6 (10 mmoles in 50 ml) was treated at -78° with 2M ethereal solution of CH₂Li (5 ml) and the mixture refluxed for 30 min. The imine⁹ rearranges to 9-acetylaminophenanthrene (of mp 197¹³) below the mp,

ir (Nujol)(C=O) 1652 cm^{-1} , uv (EtOH) 256 $\text{m}\mu$ (ϵ 40000), δ (CDCl_3) 1.92 (s, 3H), 2.30 (s, 2H), 8.10-7.35 (m, 6H), 8.66 (m, 2H). The mass spectrum provides most convincing evidence for the structure of **8**. The major peaks are 235 $[\text{M}]^+$, 193 $[\text{M}-\text{COCH}_3]^+$, and the typical fluorenyl fragment of the 9,10-dihydrophenanthrene system of $m/e = 165$.¹⁴ The absence of phenanthrene fragmentation is remarkable. While no NH proton shows up in the nmr spectrum of freshly prepared **8**, the characteristic features of an aromatic amide slowly appear when the aziridine is left in CDCl_3 for several hours at room temperature. After four days the entire aziridine has rearranged into N-acetyl-9-phenanthrylamine.¹³

Syntheses of arene imines derivatives of some carcinogenic polycyclic hydrocarbons by these two methods are now in hand.

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formation of aziridine-free 2,3,4,5-tetrahydro-1H-2-benzazepine [identified by comparison with an authentic sample, G Schroeter, A Gluschke, S Götzy, J Huang, G Irmisch, E Laves, O Schrader and G Stier, Ber dt sch chem Ges, 63, 1308 (1930)], $m/e=147$ $[M]^+$
The N-benzoyl derivative, mp 98°, $m/e=251$ $[M]^+$

- 9) All new compounds have satisfactory elementary analyses of C, H and N (within $\pm 0.3\%$ of the theoretical values)
- 10) When 1, R=COCH₃ is heated above 130° for 30 min complete isomerization to the oxazoline 4 takes place, bp 135° (1.2 mm), δ (CDCl₃) 1.88 (s, 3H), 2.02 (m, 2H), 2.60 (t, 2H), 4.47 (m, 1H), 5.44 (d, 1H, $J = 10$ cps), 7.00-7.36 (m, 4H)
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